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The change in internal energy can be found from the first law of thermodynamics: U = Q-W = (3.5 x 10 J) - (2.1 x 10 J) = 0.9 x 10 J = 90 kJ.) A gas in a cylinder is kept at a constant pressure of 3.5 x 10 Pa while 300 kJ of heat are added to it, causing the gas to expand from 0.9 m³ to 1.5 m³.

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3-5 3-23 Problem 3-22 is reconsidered. The missing properties of water are to be determined using EES, and the solution is to be repeated for refrigerant-134a, refrigerant-22, and ammonia.

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