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chemical reactions occur,
there is almost always a
change in energy, which can

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be observed as a change in temperature of the reaction mixture. Exothermic reactions release energy, resulting in an increase in temperature. Endothermic reactions absorb energy, resulting in a decrease in

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15-2 Chemical Energy and the

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Answers (cont.) • Chemists are interested in changes in energy during reactions. • Enthalpy is the heat content of a system at constant pressure. • Enthalpy (heat) of reaction is the change in enthalpy during a reaction

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